

THEORETICAL STUDY ON THE OPTICAL ACTIVITY OF SPIRO[3.3]HEPTA-1,5-DIENE

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Abstract—The spectroscopic and chiroptical properties of the title compound **1** has been predicted and interpreted by a quantum mechanical approach using the CNDO–CI modification of Del Bene and Jaffé. The calculated rotational strength of the longest wavelength transitions according to the method of Hug and Wagnière suggests left-handed chirality for levorotatory **1**. The assignment is in agreement with previous work. In addition, the same computational method has been employed for spiro[4.4]nona-1,6-diene (**2**) and spiro[4.4]nonatetraene (**3**).

The UV and CD spectra of compounds containing π electrons have been of great interest to spectroscopists and theoretical chemists. In recent years several synthetic examples of unsaturated dissymmetric spiranes have been described¹⁻³ of which one theoretical study gave rise to an assignment of absolute configuration.⁴ Two calculations† have been reported on the chirality of spiranes. Our current interest in the spectroscopic and chiroptical properties of the title compound **1**⁶ and a recent approach to the chirality of spiro[4.4]nona-1,6-diene (**2**)^{1,4} prompted us to scrutinize the surprising discrepancy in absolute configuration assignment to both spirodiene **1** and **2**. From the different kinds of approaches⁷ we chose the concept of a semiempirical wave mechanical computation of rotational strength for both spiranes. As a proper MO model for the diagnosis of the relative position of the electric and magnetic transition moments, the CNDO–CI method was preferred. The original CNDO/2 program⁹ was compared with a modified version of Del Bene and Jaffé¹⁰ concerning the reproduction of the experimental results with the electronic spectra of π -systems **1**, **2** and **3**.

In Table 1 the parameter values according to the original CNDO/2 version⁹ (method A), the Del Bene and Jaffé modification¹⁰ (method B) and the CNDO formalism of Hug and Wagnière⁴ (method C) are listed. The results of both latter calculations are compiled in Table 2. The data of method A have

been left out. This original CNDO/2 method was devised for ground state properties, not for excited states or transitions. From the results of the various computational approaches in Table 2 we may conclude that the spectroscopic data are predicted most accurately by the modified CNDO/2 approximation proposed by Del Bene and Jaffé.²⁰ This inference is in agreement with theoretical predictions for a wide variety of compounds containing π -electronic systems.¹⁰ For the oscillator and rotational strengths calculations (*vide infra*) this approximation was in fact used.

The optical activity of spirodienes **1** and **2** having C_2 symmetry can be described as the result of the interaction between both equivalent chromophoric parts of these molecules under the influence of light.⁴ It has been shown⁴ that a similar pattern for the long-wavelength part of the CD spectrum of this kind of compounds is predicted in the exciton model¹⁸ as well as with the inherently dissymmetrical concept. The same tendency is observed in the particular systems **1** and **2**. Both approaches result in a positive Cotton effect of the longest

Table 1. Parameterization (values in eV)^a

	β_{λ}^*		$\gamma_{\lambda\lambda}^*$	
	H	C	H	C
method A	-9	-21	20.40	16.06
method B	-12	-17	12.85	11.11
method C	-13.6	-20.0	12.85	10.21

^aN.B. β_{AB} and γ_{AB} are differently defined by Hug and Wagnière.

^b β_{λ}^* : empirically determined resonance integral terms.

^c $\gamma_{\lambda\lambda}^*$: one-center coulomb integrals.

^dThis value indicates $\beta(2s, 2s)$; $\beta(2p, 2p) = -10.91$.

†Two tentative determinations of the absolute configuration of spiranes whose optical activity is due only to the substitution pattern have to be mentioned. W. Kuhn and K. Bein, [Z. Phys. Chem. Abt. B, **24**, 335 (1934)], calculated the S configuration for *d*-dipyrucic erythritol and T. M. Lowry and W. C. G. Baldwin [Proc. Roy. Soc. A162, 204 (1937)], assigned the S configuration to *l*-spiro[3.3]heptane-2,6-diamine. See also reference 5.

Table 2. Singlet-singlet transitions in spirenes 1, 2 and 3 (values in nm)^a

	Method B	IP ^b	Method C ^m	IP ^b	UV (exp.)	PES (exp.)
1	[189.8(B); 182.5(A)] 200.4(B); 193.1(A) ^c	10.34 (9.68)	[158.5(B); 157.9(A)] 187.5(B); 187.7(A) ^d	11.84	~200 ^e	9.02 ^f
2	[193.5(B); 183.7(A)] ^d 204.5(B); 192.8(A) ^{cd}	10.15	— 185.4(B); 181.0(A) ^{de}	—	198 ^f	—
3	[264.1 ; 192.2] ^e 273.5 ; 205.4 ^{ef}	9.33 (8.84)	— —	—	276; 218 ^f	— 7.99 ^f

^aModels were constructed having C₂ and S₄ symmetry and the geometry of cyclobutene¹¹, cyclopentene¹² and cyclopentadiene¹³, respectively; values without CI brackets, symmetry of transition in parentheses.

^bFirst ionization potential (in eV), calculated according to equation IP = -ε and/or IP = -[E(mol) - E(mol*)]; the latter one quoted in parentheses.

^cFour π-π* configurations; if 6 HOMO and 6 LUMO States were mixed, the following values were obtained for compound 1: 206.6(b), 197.2(A).

^dPlanar model.

^eDegenerate.

^f36 Configurations.

^g120 Configurations.

^hRef. 6, ⁱRef 15, ^jRef 1a, ^kRef 14, ^lRef 24.

^mThese calculations were kindly carried out by Prof. Wagnière.

wavelength transition with B symmetry followed by a negative one of a transition with A symmetry for the planar models of 1 and 2 with a righthanded sense of screw (Fig 1). Even for the nonplanar conformers of 2 (*vide infra*) the mutual opposite pattern can be explained in both ways. We tend to view the homoconjugated 1,4 spirolinked diene systems as an inherently dissymmetric chromophore because a considerable spiroconjugation has been observed for diene 1^{6a} and because an indication has been obtained for a non-planar 1.^{6a} This latter additional feature of a distorted molecule may be explained as an endeavour of 1 to diminish the importance of a destabilizing spiroconjugative interaction.¹⁹

In order to get quantitative data concerning the optical activity and to determine the absolute configuration of spirenes 1 and 2, the rotational

strength of the transitions in the near-UV were computed by application of generally developed standard quantum mechanical methods.^{4,22} In contrast to study of Hug and Wagnière⁴ the eigenvectors of Del Bene and Jaffé's modification (*vide supra*) were used. Following a procedure of Hug and Wagnière,⁴ expressions for the integrals involving the operator $\vec{\nabla}$ and $\vec{r} \times \vec{\nabla}$ were evaluated and the rotational and dipole strengths were calculated. Not only the transition energies (*vide supra*) but also the rotational strength, dipole strength and oscillator strength¹⁶ were then refined by a configuration interaction calculation. A detailed description of CI and its mathematical formulation has been given by Pariser.²³ In this work, CI was restricted mostly to the four π-π* excitations and to a maximum quantity of 49 lowest singly excited states due to the limited amount of computer storage. In the former case the influence on the rotational strength dependent on the amount of CI was verified and it was found¹⁷ that the applied degree of mixing states was a justified selection. In Table 3 the calculations are summarized and compared with experiment. It is to be noted that for spirene 2 two additional puckered conformers are distinguished based upon model studies and on the geometry of cyclopentene. In Fig 2 both conformers having cyclopent-



Fig 1. R or 4S - spiro[3.3]hepta - 1,5 - diene (1), R or 5S - spiro[4.4]nona - 1,6 - diene (2) according to the old and new method of the Cahn, Ingold, Prelog convention,⁸ respectively, and spiro[4.4]nonatetraene (3).

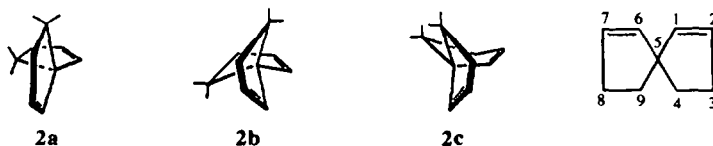


Fig 2. Geometrical isomers of R (or 5S) spiro[4.4]nona - 1,6 - diene planar 2a and puckered 2b and 2c; α = 180°, 208.8° and 151.2°, respectively.

Table 3. Comparison of experimental and calculated rotational strengths (R),^a dipole strengths (D)^b and oscillator strengths (f)

	λ^c (calc)	R (calc)	D (calc)	D ^d (exp)	f (calc)	f ^e (exp)	$\Delta\epsilon^f$ (calc)
1	200.4	-77.20	8.40	8.56	0.197	0.155	-25.3
	193.1	+82.85	1.98	—	0.048	—	+26.2
2a	204.5	-26.01	11.68	14.90	0.268	0.259	-8.7
	192.8	+29.51	0.26	—	0.006	—	+9.3
2b	208.1	-58.92	10.51	14.90	0.237	0.259	-20.1
	192.3	+52.59	1.46	—	0.036	—	+16.5
2c	204.9	+25.36	11.68	14.90	0.268	0.259	+8.5
	196.8	-10.26	0.02	—	0.001	—	-3.3
3	273.5 ^g	0	1.59	0.90	0.054	0.024	0
	205.4	0	2.73	6.36	0.063	0.115	0

^aIn units of Biot. ^bIn units of Debye. ^cIn nm.

^dCalculated according to the equation $D = 0.92 \times 10^{-38} \sqrt{\pi} \Delta / \lambda \circ \epsilon_{\max}$, where Δ is half of the bandwidth at $\epsilon = \epsilon_{\max}/E$ and is assumed to be 15 nm and $\lambda \circ$ is the wavelength at ϵ_{\max} .

^eCalculated according to the expression $f = 4.32 \times 10^{-9} \times \epsilon_{\max} \times \Delta \nu_{1/2}^2$, where $\Delta \nu_{1/2}$ is the half-width of the absorption band and is assumed to be appr. 5000 cm^{-1} .

^fCalculated according to the equation $R = 0.23 \times 10^{-38} \sqrt{\pi} \Delta / \lambda \circ \Delta \epsilon_{\max}$.

^gDegenerate.

Table 4. Results CNDO/2 calculations^a on spirodienes 2a, 2b and 2c

	E total (eV)	E el. (eV)	E n.r. (eV)
2a	0	0	0
2b	0.272	-16.00	16.28
2c	0.274	-11.08	11.36

^aMethod A.

tene geometry¹² and C₂ symmetry are represented only differing in the dihedral angle (α) of the planes C1, C2, C3, C5 and C3, C4, C5.

From Table 3 it is seen that the present work does not give conclusive evidence for assignment of absolute configuration to 2, if conformational mobility is assumed. Conformational analysis by means of this CNDO formalism (Table 4) suggests a situation which is intelligible in terms of a non-rigid system 2 containing predominantly the planar conformation. The probable equilibrium should lie substantially toward the planar form with the conformers 2b and 2c as the most puckered extremes. Contrary to this procedure Hug and Wagnière⁴ were able to assign the R or 5S configuration to planar 2 for which a positive resultant of superposed Cotton effects was calculated. On the other hand, according to the present calculation the R or 4S to levorotatory spirodiene 1 is assigned, even if a Gaussian line shape analysis is performed. This assignment is in agreement with previous work.^{6b}

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